

A FACILE CONVERSION OF AMINO TO THIONO GROUP IN CERTAIN NUCLEOBASES

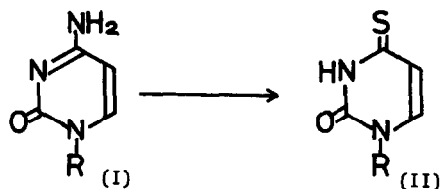
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4-Thiouridine, initially synthesized by Fox and co-workers¹, has been identified as one of the minor nucleoside components of transfer ribonucleic acid². Later, several 2-thiouridines and 2-thiocytidine have been reported as minor components^{3,4}. The replacement of a carbonyl function by a thiocarbonyl in nucleobases may provide analogs with interesting chemical and biological activities. In the course of our synthetic studies on 2-thiopyrimidine nucleosides⁵⁻⁸ we have observed a facile replacement reaction of an amino group with thiono group in cytidine derivatives by treatment with hydrogen sulfide.

To cytidine 5'-phosphate disodium salt(I, 400 mg in 5 ml of water) was added liquid hydrogen sulfide solution (H₂S gas was conducted into 5 ml of pyridine at -60°-70° to a volume of 15 ml) in a steel container and heated at 60° for 41 hours. After evaporation

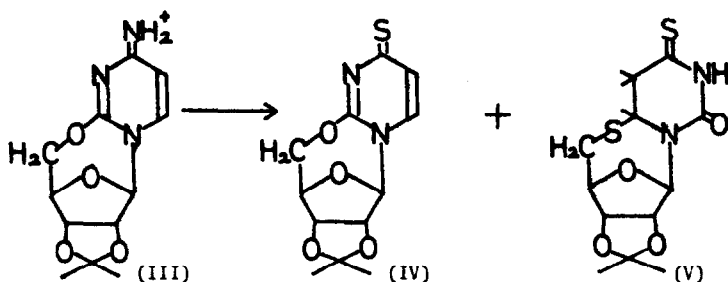


R = 5-Phosphoribosyl, Ribofuranosyl
or H.

of solvents the residue was triturated in water, filtered, and the filtrate was concentrated to a small volume from which yellow needles precipitated. Ethanol was slowly added to complete the crystallization. 4-Thiouridine 5'-phosphate⁹ disodium salt(II, 300 mg, 70%), thus obtained, showed the expected chromatographic behavior and UV(λ_{max} [m μ], ϵ : 333, 23500; 246, 4300 in 0.05 N HCl; 318, 19500 in 0.05 N NaOH) and elemental analyses(Anal. Calcd for C₉H₁₁N₂O₈Na₂PS-1.5 H₂O: C, 26.27; H, 3.41; N, 6.81. Found: C, 26.30; H, 3.90; N, 6.82). In a similar manner, cytidine(1.5g) in 50 ml of DMF with 50ml of pyridine-H₂S(1:1) at 50° for 3 days was converted into 4-thiouridine^{1,9a}(mp 135-138°, 800 mg, 50 %) after silicic acid column chromatography. 2',3'-O-Isopropylidencytidine also gave the 4-thio derivative, mp 181-184°, in 82 % yield. Cytosine

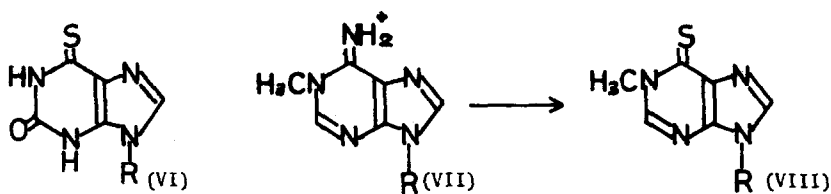
itself afforded 4-thiouracil as expected. The present method of preparation of 4-thiouracil derivative is unique when compared with the prevalent thiation procedures because of its simplicity ; prior protection of the sugar moiety is unnecessary^{1,9}.

The exchange reaction of an amino by a thiono group was also achieved with other cytidine derivatives. 2-Thiocytidine^{5,6} in H₂S-pyridine-water at 60° for 60 hours was completely converted into 2,4-dithiouridine, mp 170.5-171.5°⁵. 1-β-D-Ribofuranosyl-2,4-diaminopyrimidine⁶ also gave 2,4-dithiouridine along with an unidentified product. 3-Methylcytidine methosulfate¹⁰ was converted to 3-methyl-4-thiouridine¹¹, mp 147-148°, under very mild conditions. 2',3'-O-Isopropylidene-0²,5'-cyclocytidine p-toluenesulfonate¹² (III) gave two products depending on reaction conditions. When

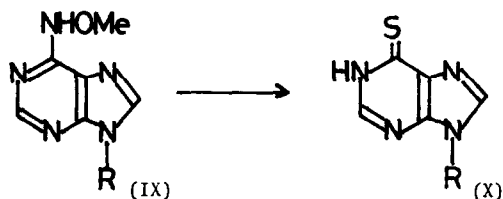


the reaction was carried out in DMF at 50° for 19 hours, 0²,5'-cyclo-4-thiouridine derivative(IV) was formed as the main product. This reaction, carried out in pyridine at room temperature for 3 days, gave 5'-deoxy-5⁶-epithio-5,6-dihydro-2',3'-O-isopropylidene-4-thiouridine(V; mp 177-179.5°, Mass spectrum, m/e; 316(M⁺); Anal. Calcd for C₁₂H₁₆N₂O₄S₂: C, 45.55; H, 5.11; N, 8.86; S, 20.25. Found: C, 45.74; H, 4.96; N, 8.89; S, 20.25). The UV spectra of V, $\lambda_{\text{max}}^{\text{EtOH}}$ at 284m μ , shifted to 320m μ on addition of sodium hydroxide solution and acidification of the solution changed the maximum to 330m μ . This absorption gradually decreased in concomitant with an increase at 285m μ . A similar phenomenon has been exemplified with 5'-deoxy-5'-mercaptouridine¹³, in which the 5'-sulfhydryl group adds at 5,6-double bond of the uracil moiety in acidic medium. Isoguanosine¹⁴, in which the pyrimidine portion is identical with the cytosine structure, was treated with H₂S-pyridine-water at 60° for 45 hours to yield 6-thioxanthosine(VI)¹⁵.

From the above results it seemed that the displacement of the amino group in cytosine derivatives would be facilitated by the quarternization(such as in 1-methyl



cytidine and III). Therefore it follows that the relatively inactive amino group in adenosine (and guanosine) could be replaced if the ring nitrogen of purine nucleus is alkylated. Treatment of 1-methyladenosine hydroiodide (VII)¹⁶ with H_2S -pyridine-water system at 60° for 46 hours afforded 1-methyl-6-thioinosine (VIII) as expected: mp $217\text{--}218^\circ$, Anal. Calcd for $\text{C}_{11}\text{H}_{14}\text{N}_4\text{O}_4\text{S}$: C, 44.30; H, 4.70; N, 18.79; S, 10.67. Found: C, 44.39; H, 4.88; N, 18.74; S, 10.67. UV $\lambda_{\text{max}}^{\text{H}_2\text{O}}$ 320, 230 μ ; $\lambda_{\text{max}}^{\text{N HCl}}$ 322, 230 μ . Adenosine itself afforded 6-thioinosine in very poor yield even under forced condition (100° , one week). However, when the leaving ability of the amino function is enhanced as in 6-methoxyadenosine (IX)¹⁷, treatment with H_2S at 50° for 57 hours afforded 6-thioinosine (X)¹⁸ in 77% yield.



This new method of preparation of pyrimidine and purine thiones should be applicable to other nitrogen-containing heterocyclic amines and studies along these lines are being undertaken. Furthermore, the treatment of ribonucleic acid with H_2S -pyridine-water at 37° for 60 hours converted most of the cytosine moieties to their 4-thiouracil counterparts without appreciable decomposition of phosphodiester linkages thus exhibiting a potential use of this method for the specific modification of nucleic acids.

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References

1. J. J. Fox, D. V. Praag, I. Wempen, I. L. Doerr, L. Cheong, J. E. Knoll, M. L. Eidinoff, A. Bendich, and G. B. Brown, J. Am. Chem. Soc., 81, 178(1959).
2. M. N. Lipsett, J. Biol. Chem., 240, 3975(1965).
3. J. Carbon, H. David, and M. H. Studier, Science, 161, 1141(1968).
4. L. Baczynskyj, K. Biemann, and R. H. Hall, Science, 151, 1481(1968).
5. T. Ueda, Y. Iida, K. Ikeda, and Y. Mizuno, Chem. Pharm. Bull. (Tokyo), 16, 1788(1968).
6. T. Ueda and H. Nishino, J. Am. Chem. Soc., 90, 1678(1968).
7. T. Ueda and H. Tanaka, Chem. Pharm. Bull. (Tokyo), 18, 1491(1970).
8. T. Ueda and S. Shibuya, Chem. Pharm. Bull. (Tokyo), 18, 1076(1970).
9. a) N. K. Kochetkov, E. I. Budowski, N. N. Shibaev, G. I. Yeliseeva, M. A. Grachev, and V. P. Demushikin, Tetrahedron, 19, 1207(1963). b) M. Saneyoshi, Chem. Pharm. Bull. (Tokyo), 19, 493(1971).
10. P. Brookes and P. D. Lawley, J. Chem. Soc., 1348(1962).
11. K. H. Scheit, Tetrahedron Letters, 113(1967).
12. V. M. Clark, A. R. Todd, and J. Zussman, J. Chem. Soc., 2952(1951).
J. Zemlicka, Collection Czech. Chem. Commun., 32 1646(1967).
13. B. Bannister and F. Kagan, J. Am. Chem. Soc., 82, 3363(1960). R. W. Chambers and V. Kurkov, J. Am. Chem. Soc., 85, 2160(1963). E. Reist, A. Benitz, and L. Goodman, J. Org. Chem., 29, 554(1964).
14. F. Cramer and G. Schlingloff, Tetrahedron Letters, 320(1964).
15. A. Yamazaki, I. Kumashiro, T. Takenishi, and M. Ikehara, Chem. Pharm. Bull. (Tokyo), 16, 2172(1968).
16. J. W. Jones and R. K. Robins, J. Am. Chem. Soc., 85, 193(1963).
17. Prepared by the adaptation of a reported method: T. Fujii, T. Itaya, C. C. Wu, and S. Yamada, Chemistry and Industry, 1967(1966).
18. J. J. Fox, I. Wempen, A. Hampton, and I. L. Doerr, J. Am. Chem. Soc., 80, 1669(1958).